

## GC-MS ANALYSIS, ANTIOXIDANT AND ANTIBACTERIAL ACTIVITY OF THE BROWN ALGAE, *PADINA TETRASTROMATICA*

Mr S.Khaja Shamshuddin, Mrs. Mala Sridevi, Mrs. Jamalpuri Jhansi  
Assistant professor<sup>1,2,3</sup>

### Department of Pharmaceutics

Global College of Pharmacy, Hyderabad. Chilkur (V), Moinabad (M), Telangana- 501504.

**ABSTRACT:** When it comes to biogenic chemicals, marine algae are a major player. They create a broad range of metabolites that have chemical activities. They have powerful phytochemical and antioxidant properties. Polysaccharides, polyunsaturated fatty acids, phlorotannins, other phenolic compounds, carotenoids, and a plethora of other metabolites and natural bioactive chemicals are all found in macroalgae, and they have antibacterial properties. Researchers have looked at the bioactive compounds found in many types of algae. This research set out to analyze the active chemicals found in the methanol extract of the brown algae *Padina tetraströmatica* for their antioxidant and antibacterial properties, as well as to identify them using gas chromatography mass spectrometry. For- we tested its antioxidant capacity, DPPH radical scavenging ability, ABTS radical cation scavenging ability, phosphomolybdenum reduction ability, and Fe<sup>3+</sup> reducing power. A maximum of 54.78% was achieved in terms of DPPH radical scavenging activity. In addition to being effective against clinical infections, the extract showed strong antibacterial action against *Shigella flexneri*, *Pseudomonas aeruginosa*, and *Vibrio cholera*. Phytochemical testing revealed the presence of phenolic chemicals, alkaloids, steroids, terpenoids, and flavonoids. The existence of fatty acids and a flavone molecule (2-Phenyl-4H-1-benzopyran-4-one), a chemical with antioxidant potential, was discovered using GC-MS analysis.

**Keywords:** *Padina tetraströmatica*, Antioxidant, Antibacterial, GC-MS, Phytochemical

## INTRODUCTION

Marine algae are abundant and potentially renewable resources that are currently being explored for their bioactive compounds<sup>1</sup>. They are rich sources of structurally diverse metabolites with great pharmaceutical and biomedical potential<sup>2,3</sup>.

The bioactive compounds from the marine algae are known to possess antibacterial, antifungal, antialgal, antiviral, antimalarial, anticancer and antifouling properties<sup>4</sup>. Many studies have proved that seaweeds are a potential source of natural antioxidants<sup>5</sup>.

The challenges in combating new diseases and resistant strains of microorganisms always demand the development of novel therapeutic agents<sup>6</sup>. The interest in marine organisms as a potential and a promising source of pharmaceutical agents has increased during recent years<sup>7</sup>. Marine macro- algae are the most interesting algae group because of their

broad spectrum of biological activities<sup>8</sup>. Research studies in the past have shown that the extracts of *Padina tetraströmatica*, have free radical scavenging ability and antibacterial activity<sup>9, 10</sup>. Also, the phytochemical composition of the algae is known to possess a wide variety of bioactive compounds<sup>11, 12</sup>.

Hence, the present study was focused at investigating the antioxidant, antibacterial activity and GC-MS analysis of active compounds present in the methanol extract of the marine brown algae, *Padina tetraströmatica*.

## MATERIALS AND METHODS:

**Collection of *Padina tetraströmatica*:** Fresh algae *Padina tetraströmatica* was measured on the basis of the scavenging activity of the stable 1, 1-diphenyl 2-picrylhydrazyl (DPPH) free radical<sup>17</sup>. One mL of 0.1 mM DPPH solution in methanol was mixed with 1 mL of various concentrations (100-600 µg/mL) of the extract. The mixture was then allowed to stand for 30 min incubation in dark. Ascorbic acid was used as the reference standard. 1mL methanol and 1 mL DPPH solution was used as the

control. The decrease in absorbance was measured using UV-Vis Spectrophotometer at 517 nm. The percentage of inhibition was calculated using the following formula:

% of DPPH radical inhibition =  $\frac{\text{Control} - \text{Sample}}{\text{Control}} \times 100$

were collected in polythene bags containing seawater from the coastal line of Mandapam

Control – Sample Control

x 100

(9°16'37.34"N; 79°7'30.78"E), Ramanathapuram district, India and transported to the laboratory. Then the algal materials were thoroughly washed in running tap water to remove the attached epiphytes and other marine debris. Finally it was washed with distilled water and allowed to dry under shade for 5

- 7 days, until the moisture was completely removed. The dried material was ground to a coarse powder using an electrical mixer and stored in air tight containers at room temperature<sup>13</sup>.

**Preparation of Crude Extract:** Finely ground algal material was extracted with methanol in the ratio of 1:10 (w/v) in a conical flask for 72 h. The mixture was filtered using filter paper in a separate container. The above process was repeated for two times with the same residue using fresh solvent. All the supernatants were collected together and then the solvent was removed by rotor evaporator.

#### **ABTS Radical Cation Scavenging Assay:**

The antioxidant capacity was estimated in terms of the ABTS [2,2'-azino-bis (3-thylbenzothiazoline - 6 - sulphonic acid)] radical cation scavenging activity

<sup>18</sup>. ABTS was obtained by reacting 7 mM ABTS stock solution with 2.45 mM potassium persulfate and the mixture was left to stand in the dark at room temperature for 12 - 16 h before use. The ABTS solution (stable for 2 days) was diluted with

5 mM phosphate-buffered saline (pH 7.4) to an absorbance of 0.70±0.02 at 730 nm. The extract in varying concentrations (10-60 µg/mL) was added to 1 mL of diluted ABTS solution, the absorbance was measured after 10 min incubation at 730 nm. The ABTS radical scavenging activity was expressed as

% of ABTS radical cation inhibition =

filtrate thus obtained was allowed to concentrate and stored for further studies<sup>14</sup>.

Control – Sample

x 100

#### **Qualitative Phytochemical Screening Procedure:**

The methanol extract of *Padina tetrastrum* was subjected to preliminary phytochemical screening using standard methods<sup>15, 16</sup>. The methanol extract was screened for different classes of phyto-constituents such as alkaloids, steroids, terpenoids, flavonoids and phenolic compounds using specific standard reagents.

#### **In vitro Antioxidant Assays:**

**DPPH Radical Scavenging Assay:** The antioxidant activity of the methanol extract of

**Phosphomolybdenum Reduction Assay:** The antioxidant capacity of the algal extract was assessed as described by Prieto *et al*<sup>19</sup>. The extract with concentrations ranging from 10 to 60 µg/mL was combined with reagent solution containing ammonium molybdate (4 mM), sodium phosphate (28 mM) and sulphuric acid (600 mM). The reaction mixture was incubated in water bath at 90°C for 90 min. The absorbance of the coloured complex was measured at 695 nm. Ascorbic acid was used as standard reference.

**Ferric (Fe<sup>3+</sup>) Reducing Power Assay:** The reducing power of the algal extract was determined by using the following procedure<sup>20</sup>. One mL extract of different concentrations (10 - 60 µg/mL) was mixed with phosphate buffer (1 mL, 0.2 M, pH 6.6) and potassium ferricyanide [K<sub>3</sub>Fe(CN)<sub>6</sub>] (1mL, 1 %). The mixtures were then incubated at 50°C for 20 min. One mL of trichloroacetic acid (10 %) and 1 mL of FeCl<sub>3</sub> (0.1 %) was added and the absorbance was measured at 700 nm using Spectrophotometer. Ascorbic acid was used as the standard reference.

#### **Antibacterial Activity of *Padina tetrastrum* on Clinical Isolates:**

The clinical isolates: *Salmonella typhi*, *Vibrio cholera*, *Shigella flexneri* and *Pseudomonas aeruginosa* were used to study the antibacterial activity by agar well-diffusion assay<sup>21</sup>. 10mg of the algal extract was dissolved in 1mL of Dimethyl sulfoxide (DMSO) to form the test extract. Fresh

overnight broth culture of the clinical pathogens were prepared and inoculated onto Mueller Hinton agar plates, by uniformly swabbing using a sterile cotton swab. Then five wells of 6mm diameter were made on the plates using a sterile cork borer. To each well the test extract in varying concentrations were added. Dimethyl sulfoxide (DMSO) was used as negative control and the broad spectrum antibiotic Tetracycline was used as positive control. The plates were incubated for 24 hours at 37 °C. After incubation, the zones of inhibition were measured. The tests were done in triplicates for quality results.

**Gas Chromatography–Mass Spectrometry (GC–MS):** For GC-MS analysis, the sample was injected into a HP-5 column (30 m X 0.25 mm i.d with 0.25 µm film thickness), Agilent technologies 6890 N JEOL GC Mate II GC-MS model. Helium was used as carrier gas with a flow rate of 1 mL/min; the injector was operated at 200 °C and column oven temperature was programmed as 50 - 250 °C at a rate of 10°C/min injection mode. The following MS conditions were used: ionization voltage of 70 eV; ion source temperature of 250 °C; interface temperature of 250 °C; mass range of 50 - 600 mass units. A chromatogram was obtained and the mass spectrum of the unknown component was compared with the spectrum of the known components stored in the NIST

library.

**Statistical Analysis:** All the experiments were conducted in triplicates and data given in tables were average of the three replicates. All data were reported as mean ± standard deviation of three replicates.

## RESULTS:

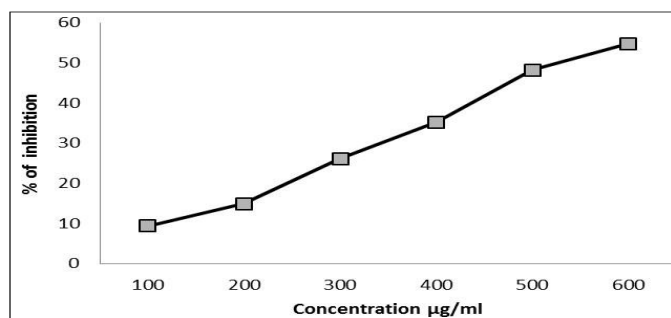
**Phytochemical Screening:** The phytochemical analysis (**Table 1**) showed the presence of alkaloids, terpenoids, steroids, phenolic compounds and flavanoids in the methanol extract of *Padinatetrasromatica*.

**TABLE 1: QUALITATIVE ANALYSIS OF PADINA TETRASTROMATICA**

S. no.	Phytochemical Constituents	Test	Result
1	Alkaloids	Mayer's test	+
2.	Terpenoids	Salkowski test	+
3.	Steroids	Liebermann's test	+
4.	Phenolic Compounds	Ferric chloride test	+
5.	Flavanoids	NaOH test	+
6.	Saponins	Foam test	-

**DPPH radical scavenging assay:** The extract demonstrated high capacity for scavenging free radicals by reducing the stable DPPH (1,1-diphenyl-2-picrylhydrazyl) radical to the yellow coloured 1,1-diphenyl-2-picrylhydrazine and the reducing capacity increased with increasing concentration of the extract.

The maximum DPPH radical scavenging activity was 54.78% at 600 µg/ml (**Table 2**). The IC<sub>50</sub> value was found to be 518.8 µg/mL concentration (**Fig.1**) and was compared with standard (Ascorbic acid, IC<sub>50</sub> =



11.98 µg/mL concentration).

**FIG. 1: DPPH RADICAL SCAVENGING ASSAY**

**TABLE 2: DPPH RADICAL SCAVENGING ACTIVITY OF *PADINA TETRASTROMATICA***

S.no.	Concentration ( $\mu\text{g/ml}$ )	% of inhibition DPPH $\cdot$
1	100	15.57 $\pm$ 1.12
2	200	28.14 $\pm$ 1.54
3	300	36.25 $\pm$ 1.77
4	400	45.13 $\pm$ 2.01
5	500	58.26 $\pm$ 2.09
6	600	61.74 $\pm$ 2.23

**ABTS Radical Cation Scavenging Assay:** The maximum ABTS radical cation scavenging activity was 57.23% at 60  $\mu\text{g/mL}$  concentration (**Table 3**). The experiment demonstrated significant antioxidant activity with  $\text{IC}_{50}$  of 53.37  $\mu\text{g/mL}$  concentration (**Fig. 2**). The results were compared with standard Ascorbic acid ( $\text{IC}_{50}$  = 4.21  $\mu\text{g/mL}$  concentration).

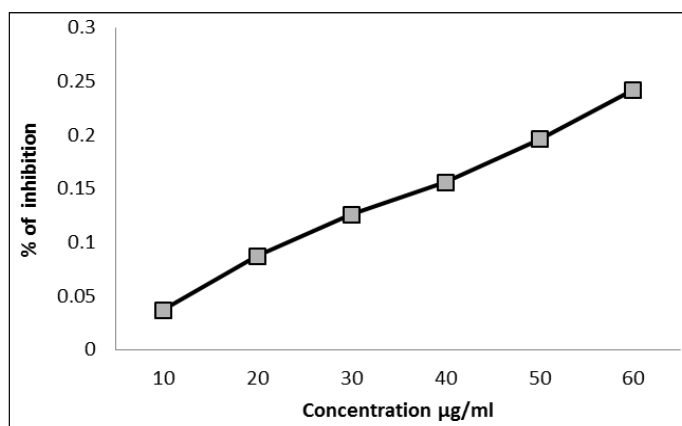
**TABLE 3: ABTS RADICAL CATION SCAVENGING ACTIVITY OF *PADINA TETRASTROMATICA***

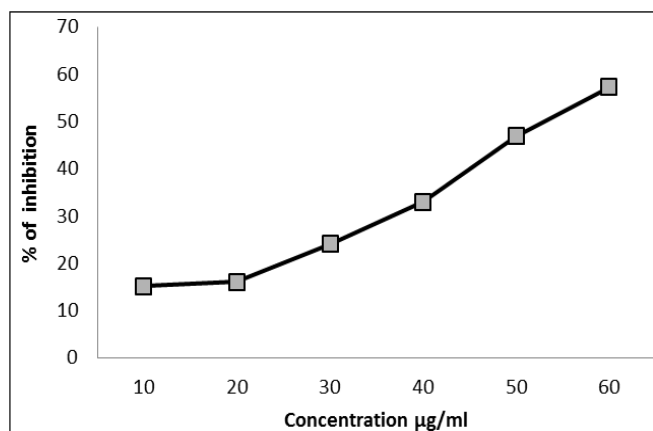
S.no.	Concentration ( $\mu\text{g/ml}$ )	% of inhibition ABTS
1	10	19.56 $\pm$ 1.11
2	20	28.96 $\pm$ 1.43
3	30	35.6 $\pm$ 1.38
4	40	48.15 $\pm$ 2.01
5	50	53.56 $\pm$ 2.29
6	60	61.26 $\pm$ 3.08

showed reduction ability in a dose dependent manner (**Fig. 4**). The maximum absorbance was 0.302 at 60  $\mu\text{g/mL}$  (**Table 4**) and was compared with the standard (0.289) Ascorbic acid.

**TABLE 4: PHOSPHOMOLYBDENUM REDUCTION AND  $\text{Fe}^{3+}$  REDUCING POWER**

S. no.	Concentration ( $\mu\text{g/ml}$ )	Phosphomolybdenum reduction assay (695 nm)	$\text{Fe}^{3+}$ reducing power assay (700 nm)
1	10	0.037 $\pm$ 0.019	0.071 $\pm$ 0.007
2	20	0.087 $\pm$ 0.014	0.123 $\pm$ 0.012
3	30	0.126 $\pm$ 0.018	0.199 $\pm$ 0.023
4	40	0.156 $\pm$ 0.011	0.237 $\pm$ 0.015
5	50	0.196 $\pm$ 0.013	0.261 $\pm$ 0.024
6	60	0.242 $\pm$ 0.021	0.302 $\pm$ 0.022

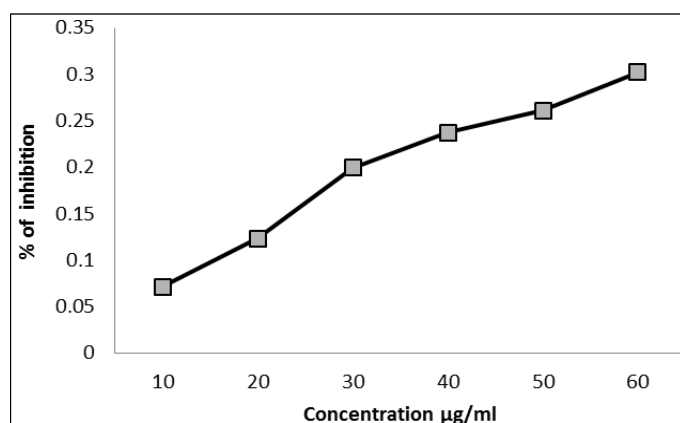
**FIG. 2: ABTS RADICAL CATION SCAVENGING ASSAY****FIG. 3: PHOSPHOMOLYBDENUM REDUCTION ASSAY**



**Phosphomolybdenum Reduction Assay Activity:** The total antioxidant activity of *Padina tetrastromatica* was measured spectro - photometrically with a maximum absorption at 695 nm (**Table 4**). The maximum absorbance was

0.242 at 60µg/mL concentration (**Fig. 3**). It was compared with the standard (0.359) Ascorbic acid.

**Ferric ( $\text{Fe}^{3+}$ ) Reducing Power Activity:** The reducing power of  $\text{Fe}^{3+}$  to  $\text{Fe}^{2+}$  was studied and



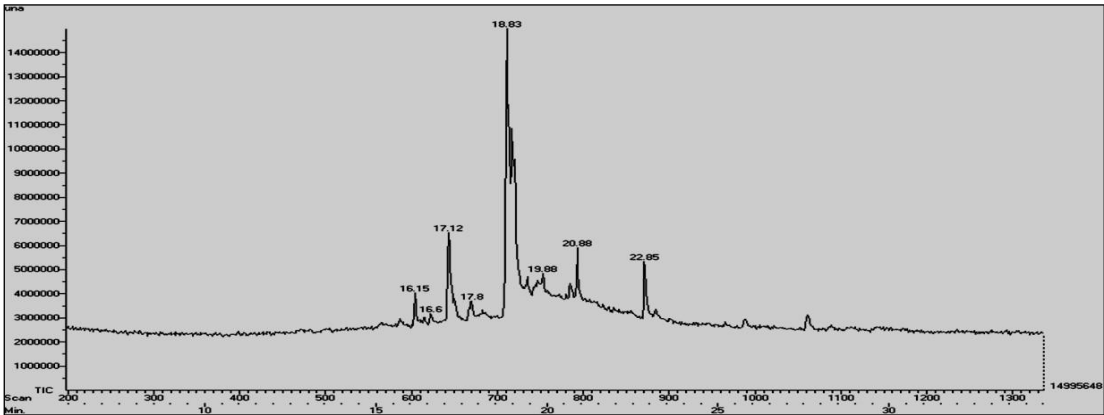
**FIG. 4: FERRIC ( $\text{Fe}^{3+}$ ) REDUCING POWER ACTIVITY**

**Antibacterial activity of *Padina tetrastromatica* on clinical isolates:** It was observed that the methanol extract of *Padina tetrastromatica* exhibited significant inhibitory activity against the clinical pathogens; *Vibrio cholera* (17mm inhibition zone), *Salmonella typhi* (15mm inhibition zone) and *Shigella flexneri* (17mm inhibition zone). However, the extract showed comparatively a lesser inhibitory activity against *Proteus mirabilis* (**Table 5**).

**TABLE 5: ANTIBACTERIAL ACTIVITY ON CLINICAL ISOLATES**

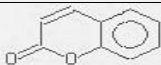
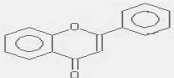
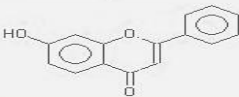
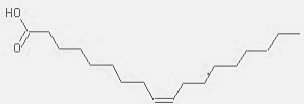

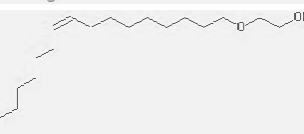
S.no.	Clinical Pathogens	Zone of inhibition(mm)					
		25µl	50µl	75µl	100µl	Standard (Tetracycline)	Negative control (DMSO)
1	<i>Salmonella typhi</i>	7	9	11	15	16	-
2	<i>Vibrio cholera</i>	9	12	13	17	21	-
3	<i>Shigella flexneri</i>	8	11	14	17	22	-
4	<i>Proteus mirabilis</i>	-	-	8	9	18	-

**GC-MS analysis:** GC-MS analysis revealed the presence of various bioactive compounds; the eluted compounds are shown in **Table 6** and the chromatogram is given below. A flavone compound (2-phenyl-4H-1-benzopyran-4-one) was eluted and it is a potential antioxidant compound.



**FIG. 5: GC SPECTRUM**

**TABLE 6: ACTIVE COMPOUNDS IDENTIFIED IN THE METHANOL EXTRACT OF *PADINA TETRASTROMATICA* BY GC - MS ANALYSIS**

S.no.	RT	Name	Structure	Mol. wt g/mol	Mol. formula	IUPAC Name
1	16.15	Coumarin		146.145	C <sub>9</sub> H <sub>6</sub> O <sub>2</sub>	2H-1-Benzopyran-2-one
2	16.6	Flavone		222.243	C <sub>15</sub> H <sub>10</sub> O <sub>2</sub>	2-Phenyl-4H-1-benzopyran-4-one
3	17.12	7-Hydroxyflavone		238.242	C <sub>15</sub> H <sub>10</sub> O <sub>3</sub>	7-hydroxy-; 4H-1-benzopyran-4-one
4	17.8	Oleic acid		282.46	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	Cis-9-Octadecenoic acid
5	18.83	Methyl oleate		296.495	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	Methyl (Z)-octadec-9-enoate
6	19.88	2-(9-Octadecenyl)oxy ethanol		312.538	C <sub>20</sub> H <sub>40</sub> O <sub>2</sub>	2-[(E)-octadec-9-enoxy]ethanol
7	20.88	Eicosanoic acid, methyl ester		326.565	C <sub>21</sub> H <sub>42</sub> O <sub>2</sub>	Methyl icosanoate
8	22.85	Methyl 5,9,13,17-tetramethyloctadecanoate		354.619	C <sub>23</sub> H <sub>46</sub> O <sub>2</sub>	Octadecanoic acid, 5,9,13,17tetramethyl, methyl ester, [5R-, (5R*,9R*,13R*)]

**DISCUSSION:** This research investigated the antioxidant activity and pathogen-inhibiting capabilities of *Padina tetrastrumatica*, an alga that has shown promise as a natural antioxidant and for the management of bacterial infections. Previous research on this algae's antibacterial properties has been published by Aseer Manilal

et al. (2016) 22 and Pushparaj et al. (2014) 23. There is strong antioxidant action, according to research by Yin Yin Chia et al. (2015). This study's findings are in line with those of Shaikh et al. (1991) 24, which also found fatty acids in *Padina tetrastrumatica* using GC-MS. The findings demonstrate that this kind of algae is a

rich reservoir of bioactive chemicals with several potential uses.

**CONCLUSION:** The present study indicates that the methanol extract of *Padina tetrastratica* have significant inhibitory activity on clinical pathogens. The phytoconstituents of the algae possess good antioxidant potential and other significant biogenic activities. The presence of 2- Phenyl-4H-1-benzopyran-4-one could be one of the reasons for the antioxidant property of the extract.

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